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IF IT'S ON THE TEST, IT'S IN THIS BOOK. The Princeton Review's MCAT® General Chemistry Review brings you everything you need to ace the gen-chem concepts found on the MCAT, including thorough subject reviews, example practice questions with step-by-step explanations, hundreds of practice problems, and 3 full-length practice tests. Inside this book, you'll find proven strategies for tackling and overcoming challenging

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General Chemistry: Understanding Moles, Bonds, and Equilibria, Volume 1 introduces students to foundational concepts in chemistry with emphasis on real-world application. Throughout the text, students learn how the study of chemistry supports material science, forensics, medicine, and other disciplines. The text is organized into 13 chapters that can be taught traditionally or in a non-linear fashion. Topics include the scientific method, atoms, mass and molecules, aqueous solutions, gases, thermochemistry, electrons in atoms, and electron configuration. Students learn about chemical bonding, molecular geometry, liquids and solids, and mixtures. The book features problems that span multiple chapters, topic boxes that contain worked examples, concurrent presentation of the VSEPR and Valence Bond theories to allow each to reinforce the other, and integration of environmental topics within distinct sections of appropriate chapters. Introductions, summaries, problems, application examples, and meaningful appendices further facilitate

student learning, rendering General Chemistry an ideal textbook for foundational chemistry courses. General Chemistry: Understanding Moles, Bonds, and Equilibria, Volume 2 is a continuation of this text with further coverage of equilibria, thermodynamics, nuclear chemistry, organic chemistry, and biochemistry and biotechnology. General chemistry is a course required for every science student regardless of career path. Within this introduction to chemistry class, students gain an essential foundation in the chemical makeup of the world around us. Whether you want to be a doctor, a nurse, earn a PhD, or enter any other field involving science in any way, gaining a strong understanding of chemistry from the start is an invaluable first step. The General Chemistry Workbook & Solutions Manual is a step-by-step guide through first-year chemistry for students who want to effectively learn chemistry while earning the best grade possible in the process. With detailed solutions to every question, this book will ensure you're ready on test day, whether that be in an undergraduate classroom or a standardized test, such as the MCAT or PCAT. It covers all of the content in standard general chemistry courses. The First Quarter is designed for students on a quarter system at their university - it includes the first third of the book, both the workbook and solutions manual, at an affordable price. If you are looking for a way to improve your science grades, earn a solid foundation in chemistry, and prepare for that next big test in your chemistry course or another test, this book is absolutely for you. The Chemistry of Ruthenium is concerned with the chemistry of ruthenium, with emphasis on synthesis and structure. The discussion spans a wide range of fields, from coordination chemistry and organometallic chemistry to structural chemistry (of both molecular and extended lattices), electrochemistry and photochemistry, as well as kinetics and spectroscopy. Comprised of 15 chapters, this book begins with an

introduction to the discovery and early history of ruthenium, along with its extraction and purification, isotopes, physical and chemical properties, and applications. The discussion then turns to the concept of oxidation state and a scheme for systematizing descriptive inorganic chemistry together with its applicability to ruthenium chemistry. Subsequent chapters focus on the chemistry of ruthenium(VIII), ruthenium(VII), ruthenium(VI), ruthenium(V), ruthenium(IV), ruthenium(III), ruthenium(II), ruthenium(I), and ruthenium(0). The book also considers ruthenium carbonyl clusters and nitrosyls before concluding with a review of the photophysics and photochemistry of tris(diimine)ruthenium(II) complexes. This monograph will be useful to students, practitioners, and researchers in the field of inorganic chemistry, as well as those who are interested in the chemistry of ruthenium. By C. Alton Hassell, Baylor University, features answers and detailed solutions to odd-numbered problems at the end of the chapter in the textbook.

A Concise and Easy Study Guide to Ace General Chemistry I Learn the important concepts covered in the first semester of a college general chemistry course in this concise but comprehensive study guide. This study guide is a supplemental resource to help students learn/review the important concepts covered in the first semester of a college general chemistry course. The guide is broken down into 11 easy to read chapters and covers: An Introduction to Chemistry Components of Matter Stoichiometry of Formulas and Equation Gases and Gas Laws Thermochemistry Quantum Theory and Atomic Structure Periodic Table and Period Properties Chemical Bonding Bonding Theories Geometry of Molecules And MUCH MUCH MORE... Buy a Copy and Begin Learning Today!

General high school chemistry text
This introductory textbook details the fundamentals of general chemistry through a wide range of topics, relating the structure of atoms and molecules to the properties of matter, in an easy to understand

format with helpful pedagogy. Ideal for chemistry courses for non-science majors, health sciences and preparatory engineering students. This General Chemistry 1 textbook starts each unit with a list of objectives, and it explores concepts that include atomic structure, chemical formulas, solution calculations, gas laws, and thermochemistry. The principles of general chemistry, stressing the underlying concepts in chemistry, relating abstract concepts to specific real-world examples, and providing a programme of problem-solving pedagogy. "Atoms First seems to be the flavor of the year in chemistry textbooks, but many of them seem to be little more than rearrangement of the chapters. It takes a master like McQuarrie to go back to the drawing board and create a logical development from smallest to largest that makes sense to students."---Hal Harris, University of Missouri-St. Louis

"McQuarrie's book is extremely well written, the order of topics is logical, and it does a great job with both introductory material and more advanced concepts. Students of all skill levels will be able to learn from this book."---Mark Kearley, Florida State University This new fourth edition of General Chemistry takes an atoms-first approach from beginning to end. In the tradition of McQuarrie's many previous works, it promises to be another ground-breaking text. This superb new book combines the clear writing and wonderful problems that have made McQuarrie famous among chemistry professors and students worldwide. Presented in an elegant design with all-new illustrations, it is available in a soft-cover edition to offer professors a fresh choice at an outstanding value. Student supplements include an online series of descriptive chemistry Interchapters, a Student Solutions Manual, and an optional state-of-the-art Online Homework program. For adopting professors, an Instructor's Manual and a CD of the art are also available. This workbook is a comprehensive collection of solved exercises and problems typical to AP, introductory, and general

chemistry courses, as well as blank worksheets containing further practice problems and questions. It contains a total of 197 learning objectives, grouped in 28 lessons, and covering the vast majority of the types of problems that a student will encounter in a typical one-year chemistry course. It also contains a fully solved, 50-question practice test, which gives students a good idea of what they might expect on an actual final exam covering the entire material. Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts. The Green Chemistry Laboratory Manual for General Chemistry provides educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual, while encouraging them to investigate the practice of green chemistry. Following a consistent format, each lab experiment begins with objectives and prelab questions highlighting important issues that must be understood prior to getting started. This is followed by detailed step-by-step procedures for performing the experiments. Students report specific results in sections designated for data, observations, and calculations. Once each experiment is completed, analysis questions test students' comprehension of the results. Additional questions encourage inquiry-based investigations and further research about how green chemistry principles compare with traditional, more hazardous experimental methods. By placing the learned concepts within the larger context of green chemistry principles, the lab manual enables students to see how these principles can be applied to real-world issues. Performing laboratory exercises through green experiments results in a safer learning environment, limits the quantity of hazardous waste generated, and reduces the cost for chemicals and waste disposal. Students using this manual will gain a greater appreciation for

green chemistry principles and the possibilities for future use in their chosen careers. This print companion to MindTap General Chemistry: Atoms First presents the narrative, figures, tables and example problems—but no graded problems or assessments. Students must use MindTap to complete the interactive activities, exercises, and assignments. The atoms first organization introduces students to atoms and molecules earlier and delays math-intensive problem-solving to later in the semester. This gives students a stronger conceptual framework to help them succeed in the course. In addition, the narrative provides greater emphasis on the historical development of the atomic nature of matter and atomic structure. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. More people get into medical school with a Kaplan MCAT course than all major courses combined. Now the same results are available with Kaplan's MCAT General Chemistry Review. This book features thorough subject review, more questions than any competitor, and the highest-yield questions available. The commentary and instruction come directly from Kaplan MCAT experts and include targeted focus on the most-tested concepts plus more questions than any other guide. Kaplan's MCAT General Chemistry Review offers:

UNPARALLELED MCAT KNOWLEDGE: The Kaplan MCAT team has spent years studying every document related to the MCAT available. In conjunction with our expert psychometricians, the Kaplan team is able to ensure the accuracy and realism of our practice materials.

THOROUGH SUBJECT REVIEW: Written by top-rated, award-winning Kaplan instructors. All material has been vetted by editors with advanced science degrees and by a medical doctor.

EXPANDED CONTENT THROUGHOUT: While the MCAT has continued to develop, this book has been updated continuously to match the AAMC's guidelines precisely—no more

worrying if your prep is comprehensive! **MORE PRACTICE THAN THE COMPETITION:** With questions throughout the book and access to one practice test, Kaplan's MCAT General Chemistry Review has more practice than any other MCAT General Chemistry book on the market. **ONLINE COMPANION:** Access to online resources to augment content studying, including one practice test. The MCAT is a computer-based test, so practicing in the same format as Test Day is key. **TOP-QUALITY IMAGES:** With full-color, 3-D illustrations, charts, graphs and diagrams from the pages of Scientific American, Kaplan's MCAT General Chemistry Review turns even the most intangible, complex science into easy-to-visualize concepts. **KAPLAN'S MCAT REPUTATION:** Kaplan gets more people into medical school than all other courses, combined. **UTILITY:** Can be used alone or with other companion books in Kaplan's MCAT Review series. Reflecting Cengage Learning's commitment to offering flexible teaching solutions and value for students and instructors, this new hybrid version features the instructional presentation found in the printed text while delivering all the end-of chapter exercises online in OWL, the leading online learning system for chemistry. The result--a briefer printed text that engages you online! Improve your grades and understanding of concepts with this value-packed Hybrid Edition of **GENERAL CHEMISTRY**, 10th edition. This cost-effective learning solution includes the textbook, the Cengage YouBook (a customizable, interactive eBook) and OWL, the leading online learning system for chemistry. To streamline the book and allow unlimited online practice, the end-of-chapter problems have been removed from the printed book, are assignable in OWL, and can be printed from the Cengage YouBook. An access code to OWL and the Cengage YouBook is included with the text, providing you with powerful online resources that include tutorials, simulations, randomized homework questions, videos, a complete interactive

electronic version of the textbook, and more! The new edition continues to offer the signature clear explanations, macro to micro orientation, enhanced problem-solving strategies, and superior art program that have made it a best-seller. Now featuring a new design and enhanced art program that convey the excitement of chemistry, this hybrid edition provides you with even more learning support through a new "Gaining Mastery Toolbox" feature in all examples, more micro-macro presentations, new two-tier questions, and a new end-of-chapter "Check List for Review." Chemistry 2e is designed to meet the scope and sequence requirements of the two-semester general chemistry course. The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The book also includes a number of innovative features, including interactive exercises and real-world applications, designed to enhance student learning. The second edition has been revised to incorporate clearer, more current, and more dynamic explanations, while maintaining the same organization as the first edition. Substantial improvements have been made in the figures, illustrations, and example exercises that support the text narrative. Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition. "An interactive presentation of general chemistry for college and university students ... While the Interactive Presentation makes up the majority of the material on these discs, other items are: ActivChemistry Software, CAChe Visualizer for Education, a tool for visualizing molecules and their properties, Interactive periodic table database, Database of Common Chemical Compounds, Plotting tool, Molecular weight and molarity calculators."--Page iii. Get a better grade in General Chemistry! Even though General Chemistry may be challenging at times; with hard work and the

right study tools, you can still get the grade you want. With David Klein's General Chemistry as a Second Language, you'll be able to better understand fundamental principles of chemistry, solve problems, and focus on what you need to know to succeed. Here's how you can get a better grade in General Chemistry: Understand the basic concepts: General Chemistry as a Second Language focuses on selected topics in General Chemistry to give you a solid foundation. By understanding these principles, you'll have a coherent framework that will help you better understand your course. Study more efficiently and effectively: General Chemistry as a Second Language provides time-saving study tips and problem-solving strategies that will help you succeed in the course. Improve your problem-solving skills: General Chemistry as a Second Language will help you develop the skills you need to solve a variety of problem types - even unfamiliar ones! Minimizes the amount of chemicals used in the lab and resultant chemical waste. Introduces new experiments designed to reduce exposure to toxic materials, lab costs and environmental pollution. Covers basic chemical concepts as well as spectroscopy and solution, physical and inorganic chemistry. Also presents several viable macroscale versions of experiments. Includes a glossary of terms as well as appendices of scientific tables and information. For courses in two-semester general chemistry. Accurate, data-driven authorship with expanded interactivity leads to greater student engagement Unrivaled problem sets, notable scientific accuracy and currency, and remarkable clarity have made Chemistry: The Central Science the leading general chemistry text for more than a decade. Trusted, innovative, and calibrated, the text increases conceptual understanding and leads to greater student success in general chemistry by building on the expertise of the dynamic author team of leading researchers and award-winning teachers. Pearson Mastering Chemistry is not included. Students, if

Mastering is a recommended/mandatory component of the course, please ask your instructor for the correct ISBN and course ID. Mastering should only be purchased when required by an instructor. Instructors, contact your Pearson rep for more information. Mastering is an online homework, tutorial, and assessment product designed to personalize learning and improve results. With a wide range of interactive, engaging, and assignable activities, students are encouraged to actively learn and retain tough course concepts. Previous ed published: 1989 Periodic table and text on lining papers Includes index and appendices. Excerpt from Introduction to General Chemistry: A Graded Course of One Hundred Lectures The Introduction to General Chemistry here with most respectfully dedicated to you, is intended to give the student a first view of the broad field of science that has been enriched by your researches, and to serve as a supplement to existing treatises of our science. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works. Interactive General Chemistry meets students where they are...with a general chemistry program designed for the way students learn. Achieve provides a new platform for Interactive General Chemistry, thoughtfully developed to engage students for better outcomes. Powerful data and analytics provide instructors with actionable insights on a platform that allows flexibility to align with a broad variety of teaching and

learning styles and the exciting Interactive General Chemistry program! Whether a student's learning path starts with problem solving or with reading, Interactive General Chemistry delivers the learning experience he or she needs to succeed in general chemistry. Built from the ground up as a digital learning program, Interactive General Chemistry combines the Sapling Learning homework platform with a robust e-book with seamlessly embedded, multimedia-rich learning resources. This flexible learning environment helps students effectively and efficiently tackle chemistry concepts and problem solving. Student-centered development In addition to Macmillan's standard rigorous peer review process, student involvement was critical to the development and design of Interactive General Chemistry. Using extensive research on student study behavior and data collection on the resources and tools that most effectively promote understanding, we crafted this complete course solution to intentionally embrace the way that students learn. Digital-first experience Interactive General Chemistry was built from the ground up to take full advantage of the digital learning environment. High-quality multimedia resources--including Sapling interactives, PhET simulations, and new whiteboard videos by Tyler DeWitt--are seamlessly integrated into a streamlined, uncluttered e-book. Embedded links provide easy and efficient navigation, enabling students to link to review material and definitions as needed. Problems drive purposeful study Our research into students' study behavior showed that students learn best by doing--so with Interactive General Chemistry, homework problems are designed to be a front door for learning. Expanding upon the acclaimed Sapling homework--where every problem contains hints, targeted feedback, and detailed step-by-step solutions--embedded resources link problems directly to the multimedia-rich e-book, providing just-in-time support at the

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some potential applications of the material into the 21st century. It includes chapters on biological effects and industrial hygiene; on synthesis, formation and manufacture; analysis, uses, environmental effects, and physical and thermodynamic properties. Reactions with organic and inorganic materials are described. Four of the seventeen chapters are devoted to a description of the carbonyl halides (especially carbonyl difluoride) related to phosgene, and a special section deals collectively with the electronic structures of carbonyl halide molecules. Featuring the first-ever comprehensive discussion of the medical effects of phosgene poisoning and the most modern methods of treating exposure victims, the book will be of interest to historians and militarists and those working in the chemical industries (heavy chemicals, agricultural and pharmaceutical), university libraries, hospitals, medical research centres, museums, environmental research centres, poison units and health and safety institutions world-wide. This laboratory manual is intended for a two-semester general chemistry course. The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life. This lab manual covers topics such as composition of compounds, reactivity, stoichiometry, limiting reactants, gas laws, calorimetry, periodic trends, molecular structure, spectroscopy, kinetics, equilibria, thermodynamics, electrochemistry, intermolecular forces, solutions, and coordination complexes.

Test Prep Books' ACS General Chemistry Study Guide: Test Prep and Practice Test Questions for the American Chemical Society General Chemistry Exam [Includes Detailed Answer Explanations] Made by Test Prep Books experts for test takers trying to achieve a great score on the ACS General Chemistry exam. This comprehensive study guide includes: Quick Overview Find out what's inside this guide! Test-Taking Strategies Learn the best tips to help overcome your exam!

Introduction Get a thorough breakdown of what the test is and what's on it! Atomic Structure Electronic Structure Formula Calculations and the Mole Stoichiometry Solutions and Aqueous Reactions Heat and Enthalpy Structure and Bonding States of Matter Kinetics Equilibrium Acids and Bases Solubility Equilibria Electrochemistry Nuclear Chemistry Practice Questions Practice makes perfect! Detailed Answer Explanations Figure out where you went wrong and how to improve! Studying can be hard. We get it. That's why we created this guide with these great features and benefits:

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discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science. This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid/base reactions which often persists through working with acid/base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshman in STEM fields. The most trusted general chemistry text in Canada is back in a thoroughly revised 11th edition. General Chemistry: Principles and Modern Applications, is the most trusted book on the market recognized for its superior problems, lucid writing, and precision of argument and precise and detailed and treatment of the subject. The 11th edition offers enhanced hallmark features, new innovations and revised discussions that

that respond to key market needs for detailed and modern treatment of organic chemistry, embracing the power of visual learning and conquering the challenges of effective problem solving and assessment. Note: You are purchasing a standalone product; MasteringChemistry does not come packaged with this content. Students, if interested in purchasing this title with MasteringChemistry, ask your instructor for the correct package ISBN and Course ID. Instructors, contact your Pearson representative for more information. If you would like to purchase both the physical text and MasteringChemistry, search for: 0134097327 / 9780134097329 General Chemistry: Principles and Modern Applications Plus MasteringChemistry with Pearson eText -- Access Card Package, 11/e Package consists of: 0132931281 / 9780132931281 General Chemistry: Principles and Modern Applications 0133387917 / 9780133387919 Study Card for General Chemistry: Principles and Modern Applications 0133387801 / 9780133387803 MasteringChemistry with Pearson eText -- Valuepack Access Card -- for General Chemistry: Principles and Modern Applications

General Chemistry for Engineers explores the key areas of chemistry needed for engineers. This book develops material from the basics to more advanced areas in a systematic fashion. As the material is presented, case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering. Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines Begins with an 'atoms first' approach, building from the simple to the more complex chemical concepts Includes engineering case studies connecting chemical principles to solving actual engineering problems Links chemistry to contemporary issues related to the interface between chemistry and engineering practices In the past two decades,

microscale techniques have soared in popularity because these techniques minimize exposure to potentially dangerous chemicals in the lab, drastically cut the amount of chemical waste, lower costs, and reduce risks of chemical fires and explosions. The result is a safer and healthier laboratory environment. Now, with *Microscale General Chemistry Laboratory with Selected Macroscale Experiments, Second Edition*, you can bring these techniques into your own chemistry lab. Thoroughly revised with updated experiments, the new Second Edition continues to offer a large variety of well-designed, easy-to-follow experiments, as well as thorough background information and an outstanding selection of questions and problems. Research in science education has recognized the importance of history and philosophy of science (HPS). Nature of science (NOS) is considered to be an essential part of HPS with important implications for teaching science. The role played by textbooks in developing students' informed conceptions of NOS has been a source of considerable interest for science educators. In some parts of the world, textbooks become the curriculum and determine to a great extent what is taught and learned in the classroom. Given this background and interest, this monograph has evaluated NOS in university level general chemistry textbooks published in U.S.A. Most textbooks in this study provided little insight with respect to the nine criteria used for evaluating NOS. Some of the textbooks, however, inevitably refer to HPS and thus provide guidelines for future textbooks. A few of the textbooks go into considerable detail to present the atomic models of Dalton, Thomson, Rutherford, Bohr and wave mechanical to illustrate the tentative nature of scientific theories --- an important NOS aspect. These results lead to the question: Are we teaching science as practiced by scientists? An answer to this question can help us to understand the importance of NOS, by providing students an HPS-based environment, so that they too

(just like the scientists) feel the thrill and excitement of discovering new things. This monograph provides students and teachers guidelines for introducing various aspects of NOS, based on historical episodes.

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